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Efficacy of limonene nano coatings on post-harvest shelf life of strawberries

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1
2 **EFFICACY OF LIMONENE NANO COATINGS ON POST-HARVEST SHELF LIFE OF**
3 **STRAWBERRIES**

4
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10
11 **ABSTRACT**

12 Strawberries are highly demanded fruits because of their color, nutritional values and appearance.
13 The aim of this study was to develop and characterize alginate and limonene liposomes as edible
14 coating materials and to determine their efficacy in shelf life extension and maintaining quality
15 parameters of ‘Chandler’ strawberries. Alginate solution (1.5% w/v) and Limonene liposomes
16 prepared from 80% lecithin and 20% PDA were used as edible coating materials. Fungal decay
17 percentage, total yeast and mold counts, headspace atmosphere analysis, total soluble solids, pH,
18 titratable acidity, total anthocyanin content and total phenolics were analyzed to assess fruit quality
19 during 14 days at 4°C of storage. Days of storage was found to be significant in maintaining the
20 quality of the strawberries. Among the coating types, limonene liposomes were found to be
21 significantly more effective in maintaining the lower concentration of carbon dioxide (CO₂), lower
22 the change in pH (3.9), and had higher total anthocyanin (43.85) content during storage than those
23 without a liposomal coating. Thus, limonene liposomes were found to be useful for extending the
24 shelf life and maintaining quality of strawberry fruits.

25 **Keywords:**

26 Strawberry, strawberry; edible coating; shelf life; limonene; liposome

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31 **1. Introduction**

32 Consumption of fruits and vegetables have gained significant global attention in recent years.
33 Fruits and vegetables are the source of nutrients such as proteins, vitamins, minerals, fibers, and
34 phytochemicals that are essential to improve human nutrition and health (Li, 2008). The
35 phytochemicals contribute to the normal functioning of the human body (Wettasinghe et al. 2002).
36 The antioxidant compounds, such as ascorbic acid (AA), lycopene, β -carotene, and phenolics
37 contribute to nutritional content of fruit and vegetables. These compounds are known to prevent
38 oxidation caused by reactive oxygen species that lead to damage the cells and DNA, and cause
39 some degenerative diseases (Hu and Jiang, 2007).

40

41 Strawberries (*Fragaria x ananassa*), with their characteristic appearance, color and nutritional
42 values, are highly consumed fruits (Almenar et al. 2007). They are considered as a good source of
43 nutrients, anthocyanins, flavonoids and phenolic compounds (Heinonen et al.1998). United states
44 is the largest producer of strawberries in the world (Wu et al. 2012) and per capita consumption
45 was 7.9 pounds in 2013 (USDA Report 2014). Of the total strawberry production, 81 percent
46 comes from fresh market (NASS 2015).

47

48 Strawberries are one of the most susceptible foods prone to physical injuries and fungal spoilage
49 (Park et al. 2005). These results in the change of several physiochemical properties, fungal growth
50 that results in short shelf life and causes a significant postharvest loss. The goal of the research is
51 to develop edible nano-coatings from plant based antimicrobials, which would maintain the
52 postharvest quality of strawberries and extend their shelf life. In our preliminary studies, edible
53 nano-coatings prepared by the nanoencapsulation of curcumin and limonene in liposomes when
54 applied on the surface of strawberries were effective in extending the shelf life and maintain the
55 quality (Dhital et al. 2017).

56

57 Application of plant based essential oils compounds as coating materials in foods have shown to
58 prevent microbial growth and loss of nutrients; and increase the shelf life of foods (Salmieri &
59 Lacroix, 2006). Essential oils of oregano, thyme, cinnamon, lemongrass and clove exhibit
60 antimicrobial activity against strains of *E. coli* (Smith-Palmer et al. 1998; Hammer et al. 1999;

61 Friedman et al. 2002). Some plant essential oils and their components are responsible for
62 increasing the sensory attributes of fruits and preventing the microbial growth. Terpene citral, a
63 citrus essential oil is known to have antimicrobial properties and contribute for sensory properties
64 of foods (Rodov et al. 1995).

65
66 Limonene ((R)-(+)-para-Mentha-1,8-diene) is obtained from essential oil of citrus fruits i.e.
67 orange, lemon, mandarin, lime, grapefruit (Moufida and Marzouk 2003). It is a colorless liquid
68 hydrocarbon regarded as safe used largely by cosmetic, food and pharmaceutical industries and
69 has Generally Recognized as Safe (GRAS) status by US Food and Drug Administration (EPA,
70 1994). Limonene has a very strong antifungal property and is effective against food spoilage fungal
71 species. It shows an antibacterial properties effective against pathogenic bacteria like;
72 *Staphylococcus aureus*, *L.monocytogenes*, *Salmonella enterica* (Sharma and Tripathi 2008;
73 Alonso-Gutierrez et al. 2013). Due to hydrophobic nature and tendency to degrade under oxidative
74 conditions, limonene possess a challenge during its application as an edible coating material
75 because of poor dispersion in water (Li and Chiang, 2012). To address this issue, low
76 concentrations of limonene are used for dispersion in water, which in turn reduces its antimicrobial
77 activity. The problem of hydrophobic nature and use of low concentrations of limonene limits its
78 efficacy while using as a coating material. A new approach of encapsulation of phytochemicals in
79 liposome that has both hydrophobic tails and hydrophilic heads through nanotechnology
80 (Umagiliyage et al. 2017).

81
82 Incorporation of antimicrobial compounds into edible films and coatings provides an innovative
83 approach to improve microbial safety and shelf life of foods (Cagri et al. 2004). Some of the most
84 commonly used antimicrobial agents in food are; benzoic acid, sodium benzoate, sorbic acid,
85 potassium sorbate and propionic acid which can be incorporated into edible films and coatings
86 (Cruz-Romero et al. 2013). Starch based edible coatings containing potassium sorbate applied on
87 the surface of fresh strawberries reduced the microbial growth and extended the shelf life (Garcia
88 et al. 1998). A bilayer edible coating made from plant based antimicrobial compounds, limonene
89 and curcumin were applied in combination with methylcellulose (MC) for improving of post-
90 harvest quality of fresh strawberries (Dhital et al. 2017). Edible films containing organic acids,

91 protein and glycerol have shown to inhibit the growth of pathogenic organisms including *L.*
92 *monocytogenes*, *S. gaminara* and *E. coli* 0157:H7 (Hettiarachchy and Satchithanandam, 2007).

93

94 Nano-technology has been extensively used to enhance the quality of fruits and vegetables (Yang
95 et al; 2010). Encapsulation of antimicrobial compounds using the approaches of nanotechnology
96 can address the problems of microbial degradation and hence improve the quality of fruits.
97 Liposomes have been used in different fields of science and technology. In food applications,
98 liposomes can be potentially used to improve the efficacy of antimicrobial compounds, delivery
99 of nutrients and protection of sensitive ingredients in foods (Lasic,1993). Liposomes are used in
100 the encapsulation of nutrients, proteins, enzymes, antimicrobial and flavors and their controlled
101 release in food environment to delay the microbial spoilage and maintain the food quality
102 (Makwana et al. 2015). Application of essential oils encapsulated in liposomes have shown to
103 improve quality and extend the shelf life of fruits and vegetables (Alikhani-Koupaei, 2014). In our
104 previous study (Dhital et al., 2017), limonene encapsulated in liposome were found to improve the
105 post-harvest shelf life of strawberries. An improved antimicrobial activity of nano-encapsulated
106 eugenol was reported by Shah et al. (2012b) against *E. coli* 0157:H7 and *Listeria monocytogenes*
107 in bovine milk. Limonene encapsulated in nano-emulsion exhibited antimicrobial activities
108 towards *Escherichia coli*, *Lactobacillus delbrueckii* and *Saccharomyces cerevisiae* (Donsi et al.
109 2012). In previous studies, liposomes polydiacetylene were used to deliver and carry antibodies
110 and drugs in mice. Results showed that injection up to 100 mg kg⁻¹ of polydiacetylene did not
111 induce acute toxicity in the mice. (Gravel, et al., 2012).

112

113 Alginate is a generic term for the salts and derivatives of alginic acid. Alginates are commercially
114 produced from brown algae *Macrocystis pyrifera*, *Laminaria hyberborea*, *Laminaria digitata*,
115 *Ascophyllum nodosum*, *Laminaria japonica*, *Edonia maxima*, *Lessonia nigrescens*, *Durvillea*
116 *Antarctica*, and *Sargassum* spp. (Draget, 2005). These compounds have good film-forming
117 properties. The alginate films are typically uniform, transparent and water-soluble. Upon addition
118 of calcium ions, alginate undergoes conformational changes resulting in the formation of calcium
119 alginate (Moe, 1995). These compounds when applied as coating materials improved the quality
120 of fruits and vegetables by reducing the shrinkage, moisture migration, oxidative rancidity, oil
121 absorption, holding volatile compounds, improvement in sensory properties of products (Hershko

122 and Nussinovitch, 1998). Alginate has wide range of application in industrial sectors due to their
123 ability to retain water, film-forming, gelling, viscosifying and stabilizing properties. Their film
124 forming properties make them useful in food processing industries. In addition, alginate coatings
125 have shown good oxygen barrier properties (Conca and Yang, 1993) that eventually can retard
126 lipid oxidation in fruits and vegetables (Kester and Fennema 1986). Alginate based coatings
127 applied to fresh cut 'Fuji' apples showed that these coatings could carry antioxidants, which are
128 responsible for the maintaining color of cut fruits during storage (Rojas-Grau et al. 2007 b)
129 Mechanical injuries due to vibration can cause in a significant loss of fruits and vegetables. A
130 study by Singh and Xu (1993) reported that about 80% of apples could be damaged by simulated
131 transportation by truck. Damage due to vibration during the transportation was noted on different
132 fruits and vegetables that include peaches, apricots, potatoes, tomatoes (Barchi et al. 2002). In
133 strawberries, In-transit vibration causes skin abrasion and bruising which makes easier for
134 microbes to enter inside the berries and cause degradation (Fischer et al. 1992).

135
136 The goal of this study was to develop novel nanocoating treatments prepared from plant-based
137 antimicrobials encapsulated in nano-liposomes. The objectives in this study were: (1) Preparation
138 and characterization of the edible coating materials, (2) application of edible coating materials on
139 strawberry fruits and analyze the quality parameter of strawberries treated with edible coatings
140 during storage and 3) compare the efficacy of coating materials based on quality parameters.

141

142 **2. Materials and methods**

143 Fresh strawberries of 'Chandler' variety purchased from local farms located in southern Illinois.
144 Berries were visually inspected for bruises, visual fungal growth, and decay. Uniform sized berries
145 were selected and stored at 4°C prior to coating application.

146

147 **2.1. Preparation of D- Limonene Liposomes**

148 Thin film dehydration method was used for the preparation of lipid film (Figure 1). Briefly, a
149 mixture of soy-based lecithin and diacetylene (PDA; 10,12-Pentacosadiynoic acid) monomer with
150 different weight ratios (100 % lecithin, 80% lecithin, 60 % and 50 % lecithin) was dissolved in 25
151 mL of dichloromethane in a 250ml round bottom flask. The solution was then subjected to rotary
152 evaporation for 1 hour to evaporate the solvent and promote bilayer film formation. The resulting

153 film was dried overnight by placing the flask on a vacuum pump followed by film hydration by
154 the addition of 50 μ M D- limonene prepared in Nano-pure water. The resulting solution then left
155 sonicated for 20 minutes. Further, the solution was placed in a probe sonicator (VCX 500, Vibra-
156 cell, Newtown, CT) at 76 °C for 15 minutes. The solution was then filtered through 0.45 μ m nylon
157 fiber to remove the lipid aggregates. Thus, obtained liposome solution was collected and kept away
158 from light at 4°C for 8 hours prior to use for experiments. Liposomes were polymerized by
159 irradiation with a UV lamp emitting at 254 nm for approximately 2-5 min using a Pen Ray (UVGL-
160 58, Minerallight, Upland, CA) UV source (4.5 mW/cm²) in air at 4°C. The polymerized liposome
161 solution was dialyzed using a membrane Spectra/Por® Biotech Cellulose Ester (CE) membrane
162 (MWCO: 100,000) for 48 hours changing the water every 4 hours. Thus obtained liposome
163 solution was collected and stored at 4°C for further studies.

164

165 **2.2. Characterization of Liposomes**

166 UV–vis absorption spectra of all the dialyzed non-polymerized and polymerized liposomes
167 prepared with varying concentration of soy-based lecithin and PDA were recorded at room
168 temperature using a PerkinElmer Lambda 25 (spectral slit width 1 nm) UV/vis spectrometer using
169 a cuvette of 1 cm path length. Sterile distilled water used as blank and to calibrate the spectrometer
170 at 400-800 nm.

171

172 **2.3. Alginate solution preparation**

173 Sodium alginate powder was dissolved in double distilled water to prepare alginate solution.
174 Briefly, Sodium alginate was dissolved in 500 ml of water upon stirring at 70 °C for 2 hours on a
175 hot plate to obtain a 1.5 % (w/v) solution.

176

177 **2.3.1. Characterization of alginate coatings**

178 **2.3.1.1 Fluorescent imaging**

179 In order to determine homogeneity of the coating, on strawberries using fluorescence
180 microscopy, we labelled alginate with amino-pyrene. Strawberries were dipped in the solution of
181 pyrene alginate and freeze- dried at -80 °C for 24 hours. In order to check the homogeneity of the
182 coating layer on strawberries, cross sectional slices of coated strawberries were made. A Leica
183 inverted fluorescent microscope was used for fluorescent alginate coated strawberries imaging. A

184 UV lamp source was used to excite the fluorescent molecules. A long pass band UV filter was
185 used to select the excitation wavelengths. The emission spectrum was collected from 420 nm to
186 500 nm.

187

188 **2.3.1.2 Scanning electron microscopy**

189 The SEM images of the alginate-coated berries were taken for the characterization of alginate
190 coatings. Briefly, alginate coatings (1.5% w/v) were applied on the berries, followed by freeze-
191 drying (-80 °C) for 24 hours. The freeze-dried samples were cut using a razor blade in small pieces
192 (5 mm by 5 mm). They were sputter coated for 4 minutes with a layer of Ag-Pd using a DESK II,
193 DENTON VACUUM sputter. The edges of the samples were grounded using a thin layer of silver
194 paint (SPI, USA). The samples were imaged using a scanning electron microscope QUANTA FEG
195 450 (FEI), the acceleration voltage was 5 kV using ETD detector at high vacuum.

196

197 **2.4. Application of coating materials**

198 The summary of application is represented in Figure 3.1. Briefly, berries were randomly selected
199 and divided into three groups depending upon coating treatment types; limonene liposome,
200 alginate and non-coated control. Each treatment was performed in triplicate and each replicate had
201 20 berries. Berries were dipped in the solutions of 50 µM liposome solutions for 10 min. For
202 Alginate coatings, 1.5 % alginate solution was cooled to room temperature and strawberries were
203 dipped in the alginate solution for 3 minutes. The berries were then immersed in 5% w/v aqueous
204 solution of CaCl₂ for 2 min. For the non-treated control samples, the berries were rinsed with sterile
205 distilled water. All the treated berries were air dried at room temperature in an UV sterilized
206 cabinet drier for 2 hours and packed in sterile clamshell box stored at 4° C.

207 **2.5 Fungal decay percentage**

208 Berries were visually evaluated for the presence of visible mold growth during the experiment.
209 Any berry with visible growth was considered to be decayed. Fungal decay percentage was
210 calculated by using the formula:

211 Fungal decay % = (the number of decayed fruits / total number of fruits) ×100

212

213 **2.6. Total Yeast and mold count**

214 Total yeast and mold count on the berry surface was performed by serial dilutions followed by
215 spread plating over the surface of sterile DRBC plate method as recommended by the International
216 Standard Organization (ISO 21527-1, 2008) with slight modifications. Briefly, berries from each
217 treatment and untreated control were stirred (in 150 mL Erlenmeyer flask) at 150 rpm in 20 mL of
218 0.1% (w/v) sterile peptone water for 30 min. The resulting suspension was then serially diluted
219 from 1:10 to 1:10⁶ dilutions. Then, 0.1 ml inoculum of each dilutions was used for plating and
220 spread evenly over the plates. The plates were incubated at 25 °C for 5 days. Results were
221 expressed as log colony forming units per ml (CFU/ml) based on average count of triplicate set.

222

223 **2.7. Headspace atmosphere analysis**

224 In hermetically sealed 500 ml glass jars, each jar containing 5 berries coated with treatments were
225 placed in and sealed. The jars were kept at 4°C for 1 hour. Head space Carbon dioxide
226 concentrations in the sealed jars were determined using an OXYBABY 6.0 gas analyzer (WITT-
227 GASETECHNIK GmbH & Co KG, Witten, Germany) comprising an electro-chemical cell for
228 oxygen analysis and an IR-absorption cell for carbon dioxide analysis. The experiment was
229 performed in triplicate.

230

231 **2.8. Fruit weight loss**

232 Strawberries just after coating and air drying were weighed. Twenty berries corresponding to each
233 coating treatment were used and the experiment was performed in triplicate. Weights of the berries
234 were measured at 2, 5, 9 and 14 days after coatings. Weight loss was estimated as the percentage
235 loss of initial weight.

236
$$\text{Weight loss \%} = (\text{Initial weight} - \text{final weight} / \text{Initial weight}) \times 100$$

237

238 **2.9. Determination of Total soluble solids (TSSs), pH, Titratable acidity (TA)**

239 TSS, TA and pH of strawberries was measured at different time intervals 2, 5, 9 and 14 days after
240 coating treatment is done. Fruit from each coating treatments were crushed with the help of sterile
241 mortar and pestle and juice will be collected. Sampling was done triplicate.

242 The TSSs of the resulting juice was measured at 20°C by a Brix refractometer (r² mini, Reichert
243 Analytical Instruments, Depew, NY). Similarly, pH of the juice was measured by a pH meter
244 (Corning pH/ion analyzer 350). TA was determined by titrating the diluted juice (5ml juice diluted
245 in 95ml distilled water) up to pH 8.2 using 0.1N NaOH.

246

247

248 **2.10. Analysis of total anthocyanin content**

249 Analysis of total anthocyanin content was performed at intervals of 2,5,9 and 14 days after coating
250 treatment is done. Strawberry sample (2g) was crushed with 20 ml of methanol in 1% HCl with
251 mortar and pestle. Then, the mixture was centrifuged at 1000×g for 20 min. The supernatant was
252 collected and absorbance was noted at 530 nm. Absorbance readings was converted to milligrams
253 of pelargonidin-3-glucoside per 100 g of fruit fresh weight, using a molar absorption coefficient
254 of 36000 (Cordenunsi et al. 2003).

255

256 **2.11. Analysis of total Phenolic compounds content**

257 Fruit samples treated with different coatings and stored at different time intervals (2, 5, 9 and 14
258 days) were selected. Briefly, a 1.5 g strawberry sample grinded in a mortar and pestle was used
259 and extracted with 20ml mixture of acetone, water and acetic acid (70:29.5:0.5 v/v). The samples
260 were vortexed for 1 hour at room temperature for complete extraction, followed by centrifugation
261 at 1640 g for 15 minutes at 20°C. The supernatant was filtered and allowed to stand at room
262 temperature for evaporation of solvent. The residue was then dissolved in distilled water to a
263 volume of 20 ml. The experiment was done in triplicate. Total phenolic content of the extracted
264 juice was determined by the use of Folin-Ciocalteu reagent as per the method of Slinkard and
265 Singleton (1977). The standard calibration curve was prepared by using Gallic acid as a standard.
266 The result was expressed as milligrams per liter of Gallic acid equivalents (GAE) per 100 gm fresh
267 weight.

268

269 **2.12. Statistical analysis**

270 The tests conducted in triplicate for each sample and simple random sampling for each tests.
271 Generalized linear mixed model analysis were carried out to determine the effect of coatings and
272 days of storage on different quality parameters. The coating treatments were compared on the basis
273 of fungal decay percentage, total yeasts and mold counts, weight loss, pH, total soluble solids
274 content and titratable acidity, total phenolic content, and total anthocyanin content. Treatment
275 means were separated using Fisher's protected least square mean separation at $P \leq 0.05$. Data were
276 analyzed using SAS 9.4 version (SAS Institute, Inc., Cary, NC).

277

278 **3. Result and discussion**

279 **3.1. Characterization of coating materials**

280 **3.1.1. Characterization of liposomes**

281 Liposomes prepared with a mixture of different concentrations of lecithin and PDA were
282 characterized by UV/Vis spectroscopy. Figure 2 shows the absorption spectrum of Lecithin: PDA
283 nanovesicles. The yellow line corresponds to unpolymerized liposomes, the lack of absorption
284 peaks demonstrates the absence of conjugation in PDA backbone. On the other hand, the blue line
285 shows the absorption spectra of liposomes prepared from the mixture of 80% lecithin and 20%
286 PDA after photo polymerization. From the absorption spectra after UV light irradiation, the peak
287 present at 655 nm along with its narrow shoulder at 593 nm corresponds to π - π^* electronic
288 transitions (Li et al. 2008; Day and Ringsdorf, 1978). The low absorbance value at lower
289 wavelengths shows low scattering that indicates low polydispersity of nanovesicles (Tomaszewska
290 et al. 2013).

291

292 **3.1.2. Characterization of alginate coatings**

293 **3.1.2.1. Fluorescence imaging of fluorescent alginate coated berries**

294 The alginate concentration for this particular experiment was roughly ten times higher than the
295 concentration of alginate originally used for the coating. As shown in the fluorescent micrographs
296 (Figure 3), the coating extends along the berry and the thickness of the coatings was about 10 μm .

297 The blue intense emission was due to the presence of amino-pyrene molecules chemically bound
298 to alginate polymer (Srivastava et al. 2009).

299

300 Alginate was labeled with amino-pyrene in order to image the presence and homogeneity of the
301 coating. Pyrene was selected due to the emission wavelength range and long lifetime excited state.
302 The emission intensity was recorded from 420 to 500 nm with a Nuance hyperspectral CCD
303 camera. Figure 4 corresponds to the fluorescent spectrum obtained from the fluorescence
304 micrograph (Fig. 3) on the amino-pyrene alginate coating. A maximum emission peak was
305 observed at 490 nm. As it is well known that pyrene emission wavelength shifts due to the presence
306 of stacking of pyrene molecules confirming the presence of self-associated pyrene excimer within
307 hydrophobic membrane of alginate coating (Uddin and Azam, 2013).

308

309 **3.1.2.2. Scanning Electron Microscopy of alginate coated berries**

310 Electron imaging (SEM) of strawberries coated with 1.5 % alginate was used to determine the
311 thickness of the coated layer. From the electron micrographs showed in Figure 5, we could
312 determine that the alginate layer was about $180 \text{ nm} \pm 40$.

313

314 **3.2 Headspace atmosphere analysis**

315 The composition of gases present in the headspace atmosphere is dependent on the physiological
316 activity of the fruits and by the microbial metabolism (Poverenov et al. 2014). There was a
317 significant change in CO_2 concentration during the storage time ($p < 0.05$) in both treated and non-
318 treated strawberries (Figure 6). Concentration of CO_2 at both 5th and 9th days of storage was
319 significantly lower than that of 2nd day, but there was a significant increase observed in the CO_2
320 concentration after 9 days of storage. The increase in CO_2 concentration at after 9 days can be
321 related with the damage in fruits and fungal decay (Hernández-Muñoz et al. 2006).

322

323 A significant difference was observed in the concentration of CO₂ in liposome treated strawberries
324 compared to those treated with alginate and non- treated control ($p < 0.05$) (Figure 7). The liposome
325 treated berries showed lower concentration of CO₂ up to 14 days of storage. These results provide
326 an evidence to the antimicrobial characteristics of limonene against spoilage microbes in fruits
327 during storage (Vu et al. 2011). The increased concentration CO₂ among alginate treated
328 strawberries can be attributed to their lower gas exchange properties (Poverenov et al. 2014).
329 Permeability of the edible coatings is one of the major factors which tend to effect the headspace
330 composition of fruits and vegetables. If the coatings is not permeable enough, normal gases
331 exchange is stopped which results in hypoxic conditions inside fruit tissue. This is indicated by
332 generation of off- flavor and enhanced production of CO₂ (Baldwin et al. 1999; Han, 2005). The
333 increase in the CO₂ composition in control and alginate treated berries can be attributed to the
334 production of CO₂, ethanol, organic acids produced by spoilage microbes (Jacxsens et al. 2003).

335

336 **3.3. Fruit weight loss**

337 The loss of weight in fruits is associated with respiration rate and evaporation of moisture through
338 the skin. The rapid loss of water from the skin is one of the major factor that contributes to the
339 perishability of strawberry fruits (Aharoni and Barkai-Golan, 1987). This leads to the dehydration
340 of fruits and ultimately to shrinkage and deterioration. Edible coatings were found to prevent water
341 transfer, protect the fruits skin from mechanical injuries resulting in delaying water loss (Ali et
342 al.2011; Chien et al. 2007). In our study, no any significant difference was observed between the
343 coating types. However, there was a significant difference noticed in between the days of storage
344 in both treated and non-treated strawberries (Figure 8) with the highest weight loss observed in 14
345 days of storage.

346

347 **3.4. pH**

348 There was a significant difference in pH of the berries between 2nd and 5th days of storage in both
349 treated and non-treated strawberries (Figure 9). The pH tend to rise significantly ($p < 0.05$) from
350 2nd to the 5th days of storage and there was a significant ($p < 0.05$) decrease in pH in the 9th day
351 compared to the 5th day, however the difference was not significant among 2nd day and 9th day.

352 Further, the pH of the berries increased on the 14th day but it was only significantly higher ($p <$
353 0.05) than 2nd day of storage. These results are in agreement with similar research conducted by
354 Holcroft and Kader (1999) who observed increase in pH with the increase in storage days. The
355 increase in pH during the storage can be related to the effects of respiration rates of fruits due to
356 the increased level of oxygen (Zheng et al. 2007).

357

358 Limonene liposome treated strawberries were found to have significantly lower pH values as
359 compared to control ($p < 0.05$) (Figure 10.). Whereas, no significant differences were found
360 between the liposome treated and alginate treated berries. Similarly, no any significant difference
361 was observed among control and alginate treated berries.

362

363 **3.5. Titratable acidity (TA)**

364 There was a non-significant difference between the coating materials. An increasing trend of TA
365 was observed up to 9th days of storage among treated (Liposome and Alginate) and untreated
366 strawberries (Figure 11). There was a significant ($p < 0.05$) increase in the TA of the strawberries
367 in 5th day of storage compared to the 2nd day. However, there was a not significant increase in the
368 TA values in 9th days compared to 5th day of storage (Figure11). Further, there was a non-
369 significant decrease in the values in the 14th days of storage. The decreased in TA content in the
370 14th day of storage can be attributed to the loss of water from fruits (Hernandez-Munoz, Almenar
371 et al. 2008) due to the respiration and microbial growth.

372

373 **3.6. Total soluble solids (TSS)**

374 There was no significant differences in the TSS level observed between the coating types. The
375 mean TSS value was tend to increase significantly ($p < 0.05$) from the 2nd days of storage to 5th
376 days of storage (Figure 12), whereas significantly ($p < 0.05$) reduced in the 9th days of storage
377 compared to 5th days. There was no significant change observed from 9th day of storage onwards.

378

379 **3.7. Total Phenolic Content (TPC)**

380 There was no significant difference on total phenolic content observed among the coating types on
381 days of storage. However, there was an increasing trend in the TPC up to 14 days of storage among
382 the treated and non-treated strawberries. There was a significant increase ($p < 0.05$) in the TPC
383 content of strawberries from 2nd day to 5th day of storage but there was no significant increase from
384 5th day onwards to the 14th day of storage. (Figure 13). These results concurred with the findings
385 by (Nunes et al. 2006). The increase in the phenolic content of strawberries during storage can be
386 attributed to the accumulation of anthocyanins and the development of its dark red-brownish color
387 (Nunes et al. 1995; Montero et al.1996).

388

389 **3.8. Total anthocyanin content**

390 Similarly, there was a significant increase in total anthocyanin content of the strawberries during
391 storage with the highest values observed in the 14th days of storage (Figure 14). These findings are
392 in agreement with the studies done by Jiang and Joyce (2003) and Ayala-Zavala et al. (2004).
393 Anthocyanins are responsible for the characteristic red color of ripe strawberries (Timberlake &
394 Bridle). They are biologically significant for their antioxidant properties (Wang et al.1996). A
395 regulatory enzyme, phenylalanine ammonia-lyase is responsible for the biosynthesis of
396 anthocyanin in fruits and vegetables (Martinez et al. 1996).

397

398 There was significant difference in total anthocyanin content of strawberries among the coating
399 types (Figure 15). The liposome treated strawberries showed significantly higher ($p < 0.05$) amount
400 of anthocyanin content compared to the alginate treated and control strawberries. Similarly.
401 Alginate treated strawberries also had significantly higher anthocyanin content compared to
402 control.

403

404 **4. Conclusion**

405 Limonene liposome was found to be an effective coating material for the shelf life extension and
406 maintaining quality parameters of the strawberries. The result obtained in this study can be helpful
407 to know that storage time significantly affects the quality of the treated and non- treated

408 strawberries. The study has shown the possibility of development and application of antimicrobial
409 phytochemicals encapsulated in liposomes. The edible coatings prepared with limonene liposomes
410 were effective in the preservation of post-harvest quality of strawberries. The strawberries coated
411 with limonene liposomes were shown to have lower respiration rates compared to control and
412 alginate coatings. Similarly, the strawberries coated with liposomes had significantly lower pH
413 values (3.9) and higher anthocyanin contents (43.849). These results suggest that limonene
414 liposomes can be effective in maintain post-harvest quality of strawberries.

415

416 **Acknowledgement**

417

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426

427 **REFERENCES:**

428

429 Alikhani-Koupaei, M. (2014). Liposomal and edible coating as control release delivery systems
430 for essential oils: comparison of application on storage life of fresh-cut banana. *Quality Assurance*
431 *and Safety of Crops & Foods*, 7(2), 175-185.

432

433 Almenar, E., Del-Valle, V., Hernández-Muñoz, P., Lagarón, J. M., Catalá, R., & Gavara, R.
434 (2007). Equilibrium modified atmosphere packaging of wild strawberries. *Journal of the Science*
435 *of Food and Agriculture*, 87(10), 1931-1939.

436 Alonso-Gutierrez, J., Chan, R., Bath, T. S., Adams, P. D., Keasling, J. D., Petzold, C. J., & Lee,
437 T. S. (2013). Metabolic engineering of *Escherichia coli* for limonene and perillyl alcohol
438 production. *Metabolic engineering*, 19, 33-41.

439 Ayala-Zavala, J. F., Wang, S. Y., Wang, C. Y., & González-Aguilar, G. A. (2004). Effect of
440 storage temperatures on antioxidant capacity and aroma compounds in strawberry fruit. *LWT-*
441 *Food Science and Technology*, 37(7), 687-695.

442 Baldwin, E. A., Burns, J. K., Kazokas, W., Brecht, J. K., Hagenmaier, R. D., Bender, R. J., &
443 Pesis, E. (1999). Effect of two edible coatings with different permeability characteristics on
444 mango (*Mangifera indica* L.) ripening during storage. *Postharvest Biology and*
445 *Technology*, 17(3), 215-226.

446 Barchi, G., Berardinelli, A., Guarnieri, A., Ragni, L., & Fila, C. T. (2002). PH—postharvest
447 technology: damage to loquats by vibration-simulating intra-state transport. *Biosystems*
448 *Engineering*, 82(3), 305-312.

449 Cagri, A., Ustunol, Z., & Ryser, E. T. (2004). Antimicrobial edible films and coatings. *Journal of*
450 *food protection*, 67(4), 833-848.

451 Conca, K., & Yang, T. (1993). Edible food barrier coatings. Activities report of the R and D
452 Associates (USA).

453 Cordenunsi, B. R., Nascimento, J. D., & Lajolo, F. M. (2003). Physico-chemical changes related
454 to quality of five strawberry fruit cultivars during cool-storage. *Food Chemistry*, 83(2), 167-173.

455 Cruz-Romero, M. C., Murphy, T., Morris, M., Cummins, E., & Kerry, J. P. (2013).
456 Antimicrobial activity of chitosan, organic acids and nano-sized solubilisates for potential use in
457 smart antimicrobially-active packaging for potential food applications. *Food Control*, 34(2), 393-
458 397

459 Day, D., & Ringsdorf, H. (1978). Polymerization of diacetylene carbonic acid monolayers at the
460 gas-water interface. *Journal of Polymer Science Part C: Polymer Letters*, 16(5), 205-210.

461 Dhital, R., Joshi, P., Becerra-Mora, N., Umagiliyage, A., Chai, T., Kohli, P., & Choudhary, R.
462 (2017). Integrity of edible nano-coatings and its effects on quality of strawberries subjected to
463 simulated in-transit vibrations. *LWT-Food Science and Technology*, 80, 257-264.

464 Donsì, F., Annunziata, M., Vincensi, M., & Ferrari, G. (2012). Design of nanoemulsion-based
465 delivery systems of natural antimicrobials: effect of the emulsifier. *Journal of biotechnology*,
466 159(4), 342-350.

467 Draget, K. I., Smidsrød, O., & Skjåk-Bræk, G. (2005). Alginates from algae. *Biopolymers*
468 *Online*.

469 EPA, U. (1994). *Reregistration Eligibility Decision (RED) Limonene*. (EPA 738-R-94-034).
470 Washington, DC, USA

471 Fischer, D., Craig, W., Watada, A., Douglas, W., & Ashby, B. (1992). Simulated in-transit
472 vibration damage to packaged fresh market grapes and strawberries. *Applied engineering in*
473 *agriculture*, 8(3), 363-366.

474 Friedman, M., Henika, P. R., & Mandrell, R. E. (2002). Bactericidal activities of plant essential
475 oils and some of their isolated constituents against *Campylobacter jejuni*, *Escherichia coli*,
476 *Listeria monocytogenes*, and *Salmonella enterica*. *Journal of food protection*, 65(10), 1545-1560.

477 Garcia, M. A., Martino, M. N., & Zaritzky, N. E. (1998). Plasticized starch-based coatings to
478 improve strawberry (*Fragaria* × *ananassa*) quality and stability. *Journal of Agricultural and*
479 *Food Chemistry*, 46(9), 3758-3767.

480 Gravel, E., Ogier, J., Arnauld, T., Mackiewicz, N., Ducongé F., Doris, E. Drug Delivery and
481 Imaging with Polydiacetylene Micelles. *Chem. Eur. L.* 2012, 18, 400-408.

482 Hammer, K. A., Carson, C., & Riley, T. (1999). Antimicrobial activity of essential oils and other
483 plant extracts. *Journal of applied microbiology*, 86(6), 985-990.

484 Han, J. H. (Ed.). (2005). *Innovations in food packaging*. Academic Press.

485 Heinonen, I. M., Meyer, A. S., & Frankel, E. N. (1998). Antioxidant activity of berry phenolics
486 on human low-density lipoprotein and liposome oxidation. *Journal of Agricultural and Food*
487 *Chemistry*, 46, 4107-4112

488 Hernández-Muñoz, P., Almenar, E., Ocio, M. J., & Gavara, R. (2006). Effect of calcium dips and
489 chitosan coatings on postharvest life of strawberries (*Fragaria x ananassa*). *Postharvest Biology*
490 *and Technology*, 39(3), 247-253.

491 Hershko, V., & Nussinovitch, A. (1998). Physical properties of alginate-coated onion (*Allium*
492 *cepa*) skin. *Food Hydrocolloids*, 12(2), 195-202.

493

494 Hettiarachchy, N. S., & Satchithanandam, E. (2007). Organic acids incorporated edible
495 antimicrobial films: Google Patents

496 Holcroft, D. M., & Kader, A. A. (1999). Controlled atmosphere-induced changes in pH and
497 organic acid metabolism may affect color of stored strawberry fruit. *Postharvest Biology and*
498 *Technology*, 17(1), 19-32.

499 Hu, W., & Jiang, Y. (2007). Quality attributes and control of fresh-cut produce. *Stewart*
500 *Postharvest Rev*, 3, 1-9.

501 Jacxsens, L., Devlieghere, F., Ragaert, P., Vanneste, E., & Debevere, J. (2003). Relation
502 between microbiological quality, metabolite production and sensory quality of equilibrium
503 modified atmosphere packaged fresh-cut produce. *International Journal of Food*
504 *Microbiology*, 83(3), 263-280.

505 Jiang, Y., & Joyce, D. C. (2003). ABA effects on ethylene production, PAL activity, anthocyanin
506 and phenolic contents of strawberry fruit. *Plant Growth Regulation*, 39(2), 171-174.

507 Lasic, D.D. (1993). *Liposomes from Physics to Applications*; Elsevier: Amsterdam, 507–516.

508 Li, T. S. (2008). *Vegetables and fruits: nutritional and therapeutic values*: CRC Press.

509 Li, X., Matthews, S., & Kohli, P. (2008). Fluorescence resonance energy transfer in
510 polydiacetylene liposomes. *The journal of physical chemistry. B*, 112(42), 13263.

511 Makwana, S., Choudhary, R., Haddock, J., & Kohli, P. (2015). In-vitro antibacterial activity of
512 plant based phenolic compounds for food safety and preservation. *LWT-Food Science and*
513 *Technology*, 62(2), 935-939.

514 Martinez, G. A., Chaves, A. R., & Anon, M. C. (1996). Effect of exogenous application of
515 gibberellic acid on color change and phenylalanine ammonia-lyase, chlorophyllase, and
516 peroxidase activities during ripening of strawberry fruit (*Fragaria x ananassa* Duch.). *Journal of*
517 *Plant Growth Regulation*, 15(3), 139-146.

518 Moe, S. (1995). Superswelling alginate gels: Preparation and some physical properties.

519 Montero, T. M., Mollá, E. M., Esteban, R. M., & López-Andréu, F. J. (1996). Quality attributes
520 of strawberry during ripening. *Scientia Horticulturae*, 65(4), 239-250.

521 Moufida, S. d., & Marzouk, B. (2003). Biochemical characterization of blood orange, sweet
522 orange, lemon, bergamot and bitter orange. *Phytochemistry*, 62(8), 1283-1289.

523 Noncitrus Fruits and Nuts Summary, (2015) USDA National Agricultural Statistics Service
524 (NASS).

525 Nunes, M. C. N., Brecht, J. K., Morais, A. M. M. B., & Sargent, S. A. (1995). Physical and
526 chemical quality characteristics of strawberries after storage are reduced by a short delay to
527 cooling. *Postharvest Biology and Technology*, 6(1-2), 17-28.

528 Nunes, M. C. N., Brecht, J. K., Morais, A. M., & Sargent, S. A. (2006). Physicochemical
529 changes during strawberry development in the field compared with those that occur in harvested
530 fruit during storage. *Journal of the Science of Food and Agriculture*, 86(2), 180-190.

531 Park, S. I., Stan, S. D., Daeschel, M. A., & Zhao, Y. (2005). Antifungal coatings on fresh
532 strawberries (*Fragaria × ananassa*) to control mold growth during cold storage. *Journal of food*
533 *science*, 70(4).

534 Poverenov, E., Danino, S., Horev, B., Granit, R., Vinokur, Y., & Rodov, V. (2014). Layer-by-
535 layer electrostatic deposition of edible coating on fresh cut melon model: anticipated and
536 unexpected effects of alginate–chitosan combination. *Food and bioprocess technology*, 7(5),
537 1424-1432.

538 Rodov, V., Ben-Yehoshua, S., Fang, D. Q., Kim, J. J., & Ashkenazi, R. (1995). Preformed
539 antifungal compounds of lemon fruit: citral and its relation to disease resistance. *Journal of*
540 *Agricultural and Food Chemistry*, 43(4), 1057-1061.

541 Rojas-Grau, M. A., Avena-Bustillos, R. J., Olsen, C., Friedman, M., Henika, P. R., Martin-
542 Belloso, O., et al. (2007). Effects of plant essential oils and oil compounds on mechanical,
543 Barrier and antimicrobial properties of alginate – Apple puree edible films. *Journal of Food*
544 *Engineering*, 81(3), 634–641.

545 Salmieri, S., & Lacroix, M. (2006). Physicochemical properties of alginate/polycaprolactone-
546 based films containing essential oils. *Journal of Agricultural and Food Chemistry*, 54(26),
547 10205-10214.

548 Shah, B., Davidson, P. M., & Zhong, Q. (2013). Nanodispersed eugenol has improved
549 antimicrobial activity against *Escherichia coli* O157: H7 and *Listeria monocytogenes* in bovine
550 milk. *International journal of food microbiology*, 161(1), 53-59.

551 Sharma, N., & Tripathi, A. (2008). Effects of *Citrus sinensis* (L.) Osbeck epicarp essential oil on
552 growth and morphogenesis of *Aspergillus niger* (L.) Van Tieghem. *Microbiological Research*,
553 163, 337-344.

554 Singh S P; Xu M (1993). Bruising in apples as a function of truck vibration and packaging.
555 *Applied Engineering in Agriculture*, 9(5), 455–460.

556 Slinkard, K., & Singleton, V. L. (1977). Total Phenol Analysis - Automation and Comparison
557 with Manual Methods. *American Journal of Enology and Viticulture*, 28, 49-55.

558 Smith-Palmer, A., Stewart, J., & Fyfe, L. (1998). Antimicrobial properties of plant essential oils
559 and essences against five important food-borne pathogens. *Letters in applied microbiology*,
560 26(2), 118-122.

561 Srivastava, A., Waite, J. H., Stucky, G. D., & Mikhailovsky, A. (2009). Fluorescence
562 investigations into complex coacervation between polyvinylimidazole and sodium
563 alginate. *Macromolecules*, *42*(6), 2168.

564 Tomaszewska, E., Soliwoda, K., Kadziola, K., Tkacz-Szczesna, B., Celichowski, G., Cichomski,
565 M., & Grobelny, J. (2013). Detection limits of DLS and UV-Vis spectroscopy in characterization
566 of polydisperse nanoparticles colloids. *Journal of Nanomaterials*, *2013*, 60.

567 U.S. Strawberry Consumption Continues to Grow, USDA Economic Research Service (ERS),
568 2014.

569 Volkov, V. V., Asahi, T., Masuhara, H., Masuhara, A., Kasai, H., Oikawa, H., & Nakanishi, H.
570 (2004). Size-dependent optical properties of polydiacetylene nanocrystal. *The Journal of*
571 *Physical Chemistry B*, *108*(23), 7674-7680.

572 Vu, K. D., Hollingsworth, R. G., Leroux, E., Salmieri, S., & Lacroix, M. (2011). Development of
573 edible bioactive coating based on modified chitosan for increasing the shelf life of
574 strawberries. *Food Research International*, *44*(1), 198-203.

575 Wang, H., Cao, G., & Prior, R. L. (1996). Total antioxidant capacity of fruits. *Journal of*
576 *Agricultural and Food Chemistry*, *44*(3), 701-705.

577 Wettasinghe, M., Bolling, B., Plhak, L., & Parkin, K. (2002). Screening for Phase II Enzyme-
578 inducing and Antioxidant Activities of Common Vegetables. *Journal of food science*, *67*(7),
579 2583-2588.

580 Yang, F., Li, H., Li, F., Xin, Z., Zhao, L., Zheng, Y., & Hu, Q. (2010). Effect of Nano-Packing
581 on Preservation Quality of Fresh Strawberry (*Fragaria ananassa* Duch. cv *Fengxiang*) during
582 Storage at 4° C. *Journal of food science*, *75*(3).

583 Zheng, Y., Wang, S. Y., Wang, C. Y., & Zheng, W. (2007). Changes in strawberry phenolics,
584 anthocyanins, and antioxidant capacity in response to high oxygen treatments. *LWT-Food*
585 *Science and Technology*, *40*(1), 49-57.

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Figure Caption:

Figure 1: Flowchart for the coating treatments

Figure 2: Absorption spectrum of polymerized and non-polymerized 80% lecithin 20% Polydiacetylene liposomes

Figure 3: Fluorescence micrographs of strawberries coated with pyrene alginate (scale bar 100 μm). Blue emission comes from amino-pyrene excited state.

Figure 4: Emission spectrum of pyrene alginate coating. The emission corresponds to multiple points enclosed in Figure 3.

Figure 5: SEM images alginate coated strawberries (cross section). A) Uncoated strawberry (scale bar 5 μm). B). Alginate coated strawberry (scale bar 3 μm) C) Zoomed area showing alginate layer on the alginate coated strawberry.

Figure 6: Concentration of CO_2 on various days of storage in both treated and non-treated strawberries. LS-means with the same letter are not significantly different

Figure 7: Concentration of CO_2 in strawberries with various coatings up to 14 days of storage. LS-means with the same letter are not significantly different

Figure 8: Percentage weight loss on days of storage in both treated and non-treated strawberries . LS-means with the same letter are not significantly different

Figure 9: Mean pH on days of storage in both treated and non-treated strawberries. LS-means with the same letter are not significantly different

Figure 10: Mean pH on coating types. LS-means with the same letter are not significantly different

624 Figure 11: Mean Titratable acidity on days of storage in both treated and non-treated strawberries.
625 LS-means with the same letter are not significantly different

626 Figure 12: Mean TSS on days of storage in both treated and non-treated strawberries. LS-means
627 with the same letter are not significantly different

628 Figure 13: Mean TPC on days of storage in both treated and non-treated strawberries . LS-means
629 with the same letter are not significantly different

630 Figure 14: Mean Total anthocyanin on days of storage. LS-means with the same letter are not
631 significantly different

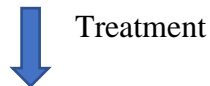
632 Figure 15: Mean Total anthocyanin on types of coatings. LS-means with the same letter are not
633 significantly different

634

635

Strawberries

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638 1) Limonene Liposome (50 μ M) 2) Alginate (1.5 % w/v) 3) Control

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640

Air-dried for 2 hours under the hood

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Packed in sterile clamshell box and stored at 4° C

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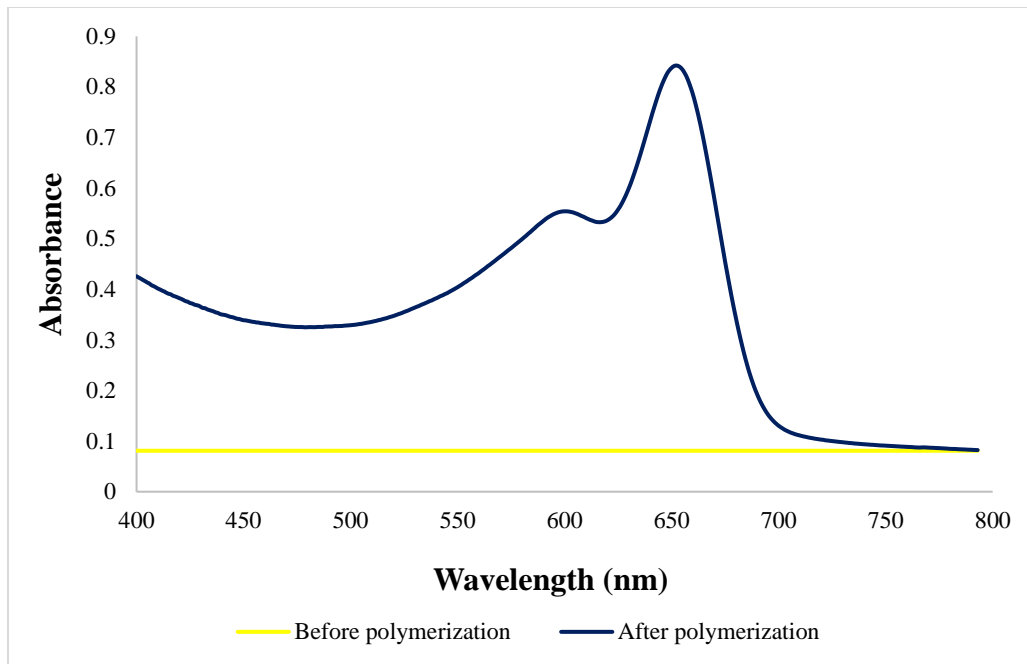


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Storage studies on different time intervals

645 Figure 1: Flowchart for the coating treatments

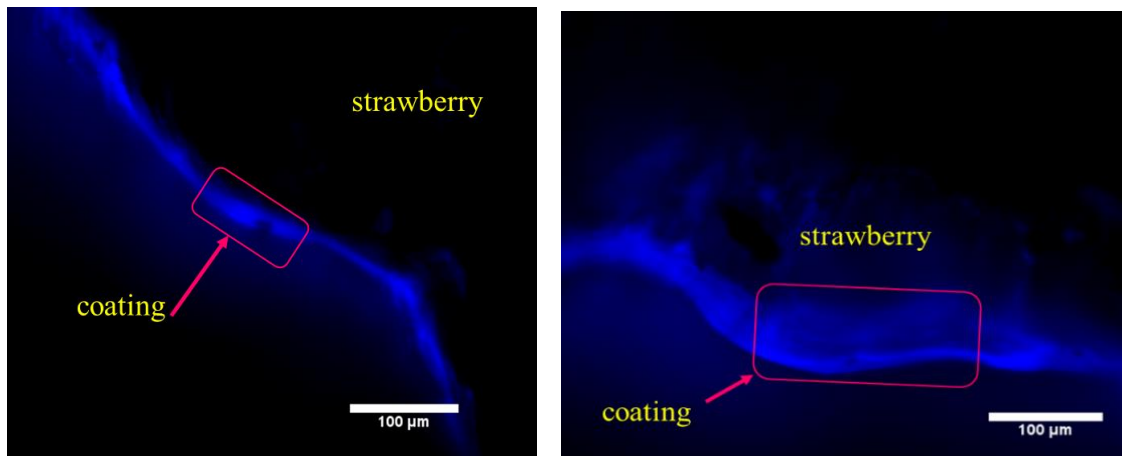
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648 Figure 2: Absorption spectrum of polymerized and non-polymerized 80% lecithin 20%
 649 Polydiacetylene liposomes

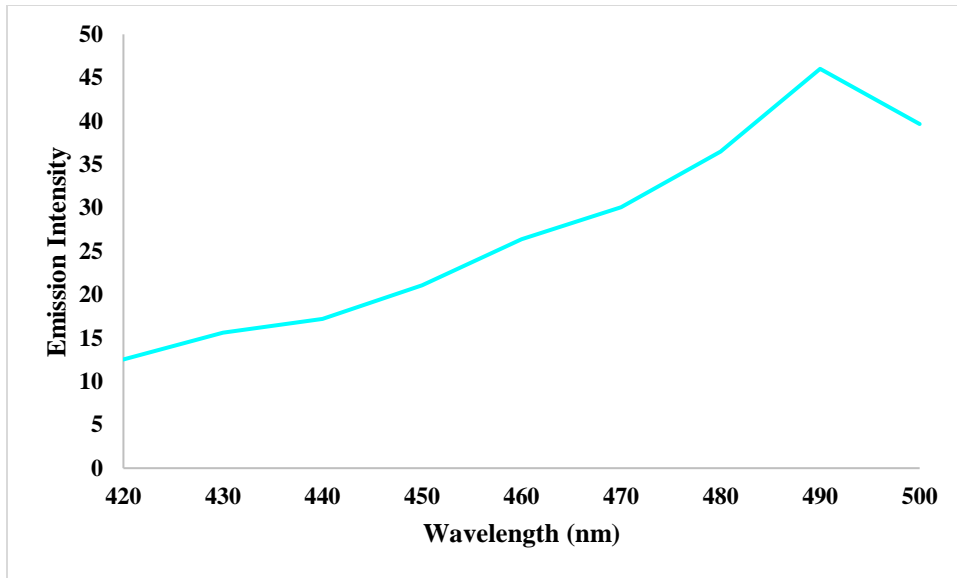
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652 Figure 3: Fluorescence micrographs of strawberries coated with pyrene alginate (scale bar 100
 653 μm). Blue emission comes from amino-pyrene excited state.

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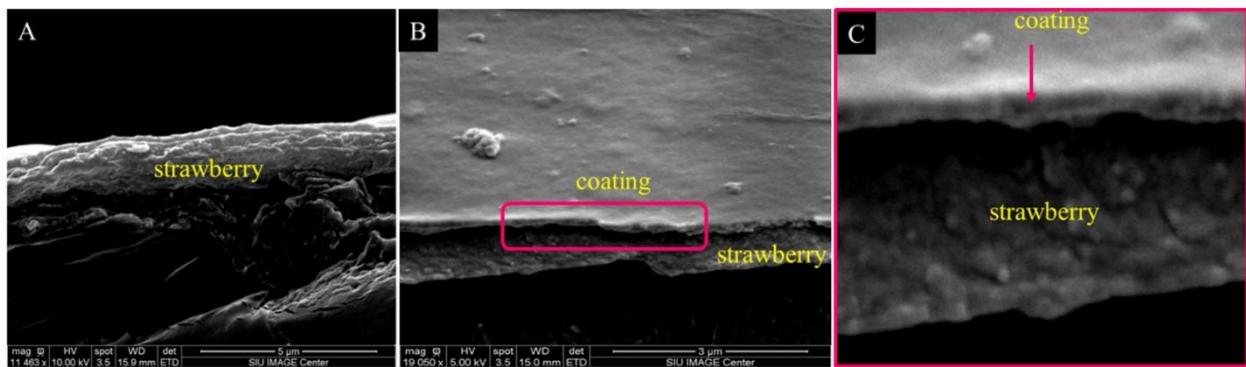
656 Figure 4: Emission spectrum of pyrene alginate coating. The emission corresponds to multiple
 657 points enclosed in Figure 3.

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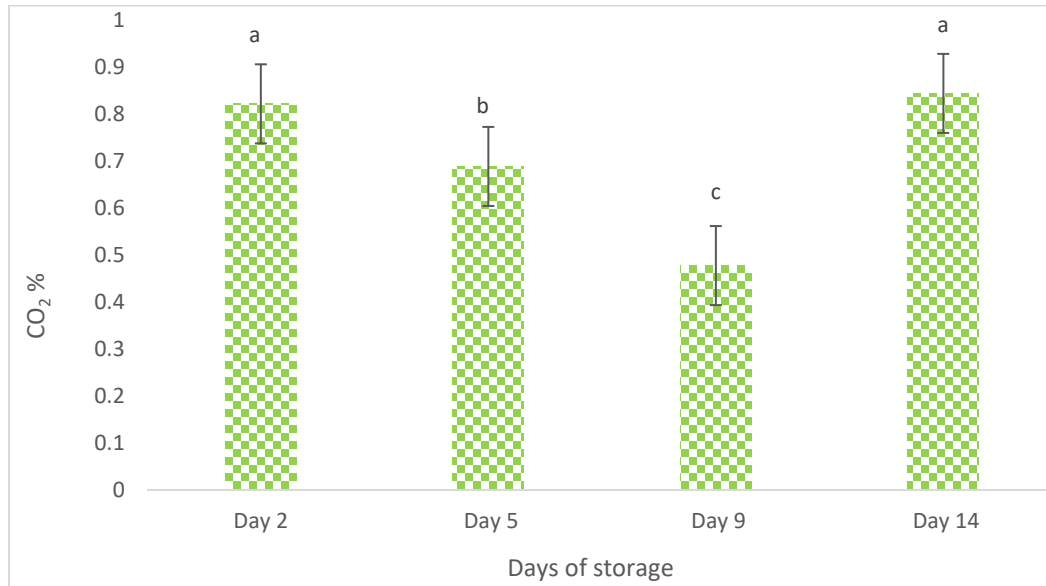


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663 Figure 5: SEM images alginated strawberries (cross section). A) Uncoated strawberry (scale
 664 bar 5 μ m). B). Alginated strawberry (scale bar 3 μ m) C) Zoomed area showing alginated layer
 665 on the alginated strawberry.

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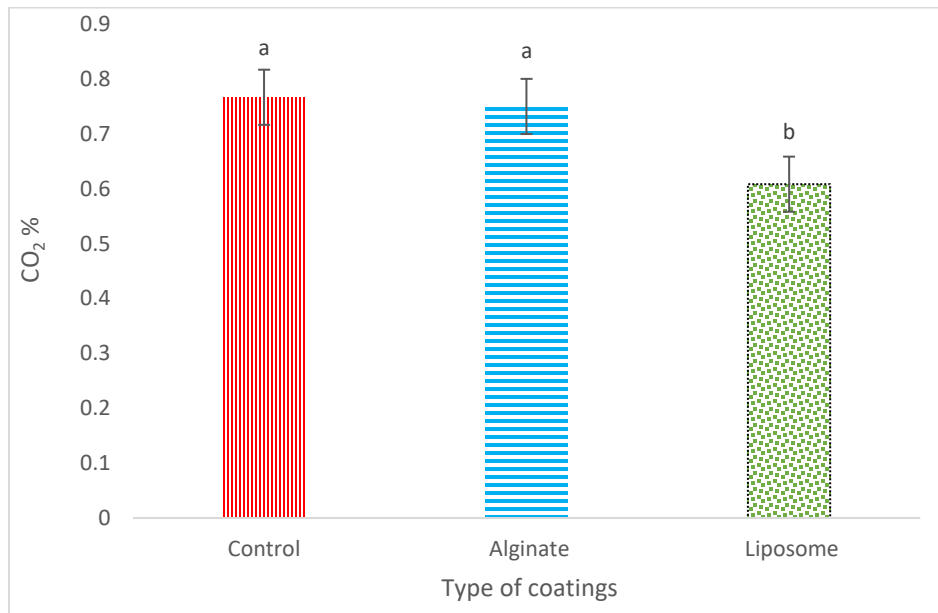
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669 Figure 6: Concentration of CO₂ on various days of storage in both treated and non-treated
670 strawberries. LS-means with the same letter are not significantly different

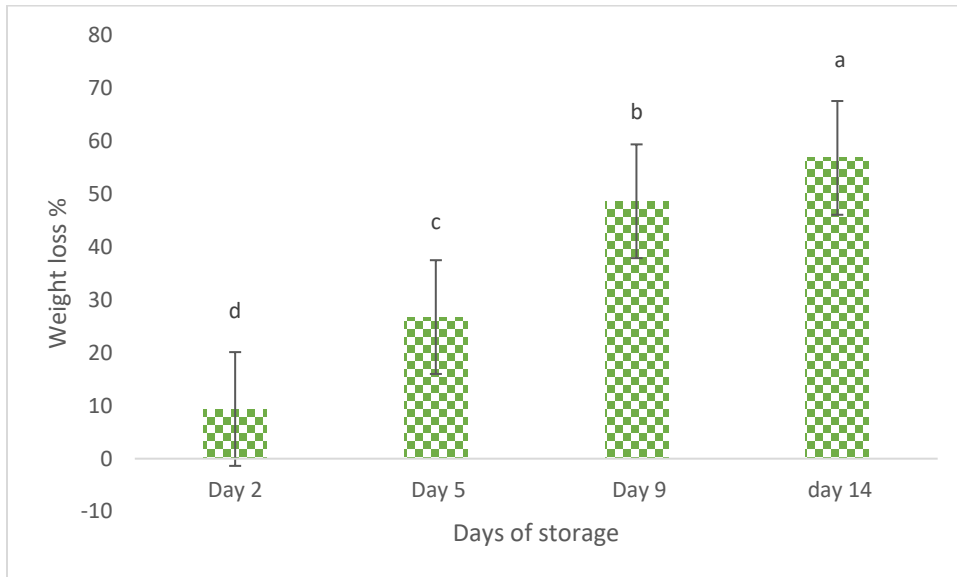
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673 Figure 7: Concentration of CO₂ in strawberries with various coatings up to 14 days of storage. LS-
674 means with the same letter are not significantly different

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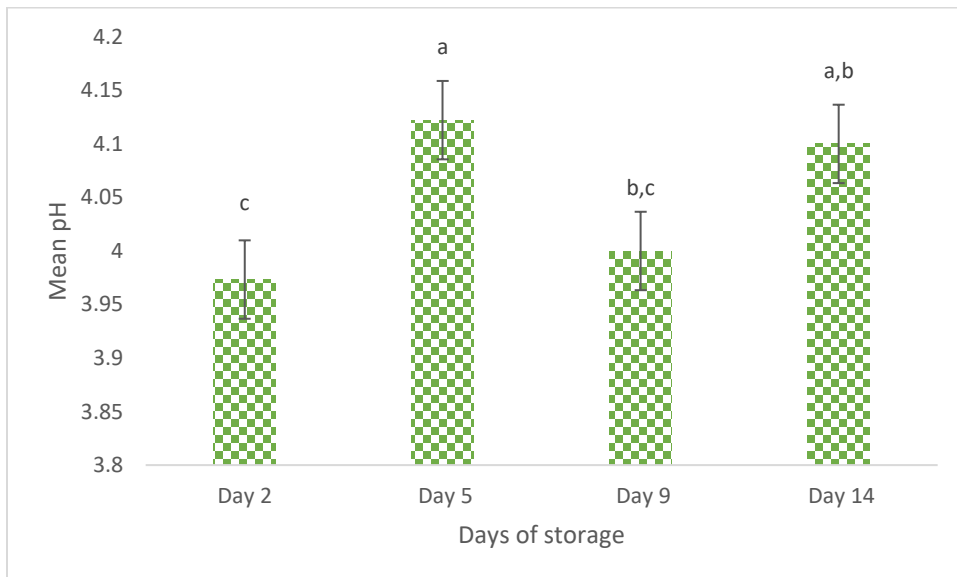


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677 Figure 8: Percentage weight loss on days of storage in both treated and non-treated strawberries .

678 LS-means with the same letter are not significantly different

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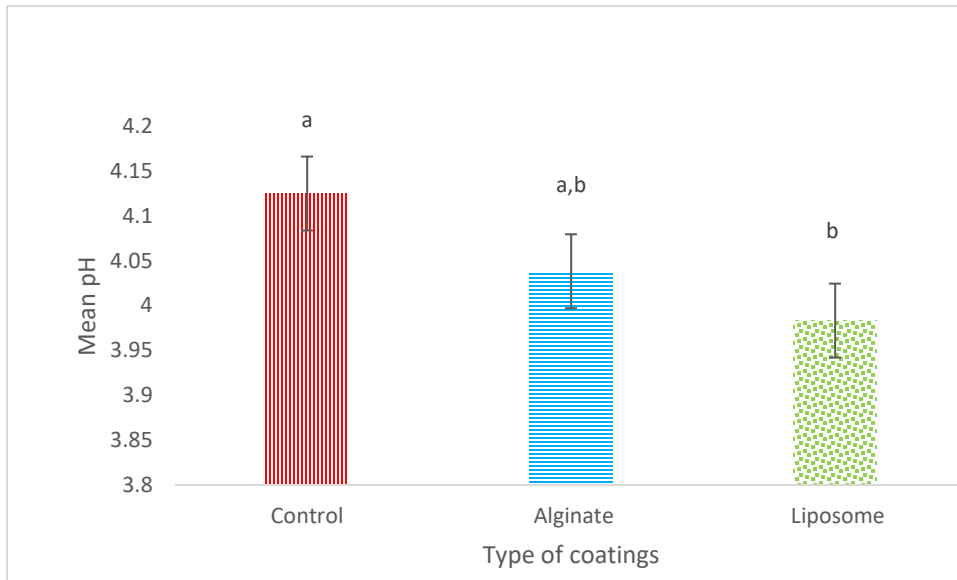
681 Figure 9: Mean pH on days of storage in both treated and non-treated strawberries. LS-means with

682 the same letter are not significantly different

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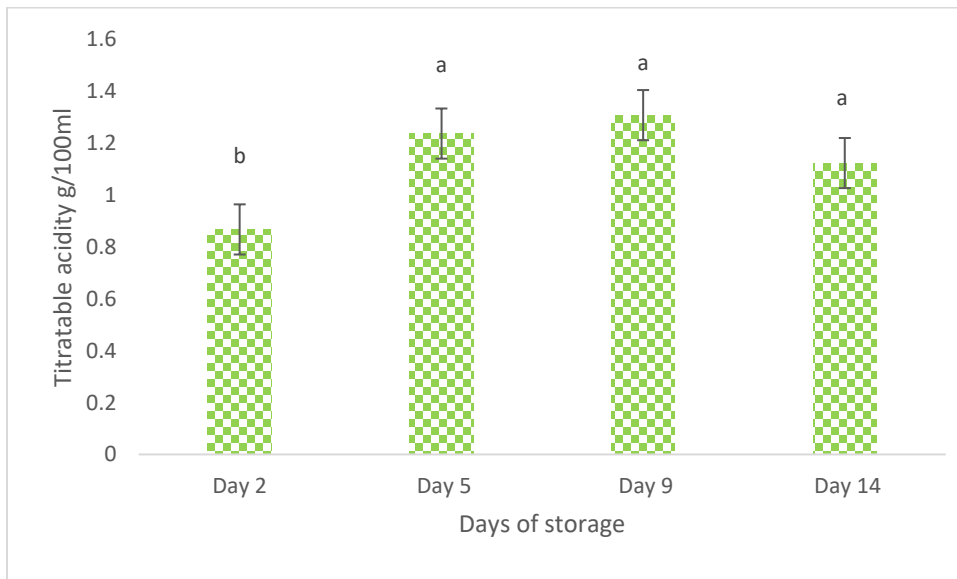
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687 Figure 10: Mean pH on coating types. LS-means with the same letter are not significantly different

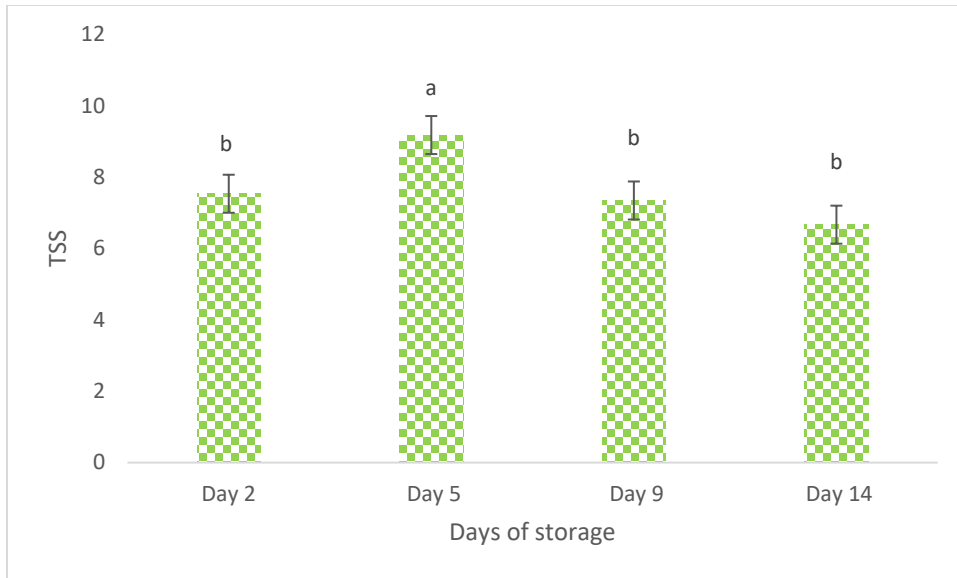
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690 Figure 11: Mean Titratable acidity on days of storage in both treated and non-treated strawberries.

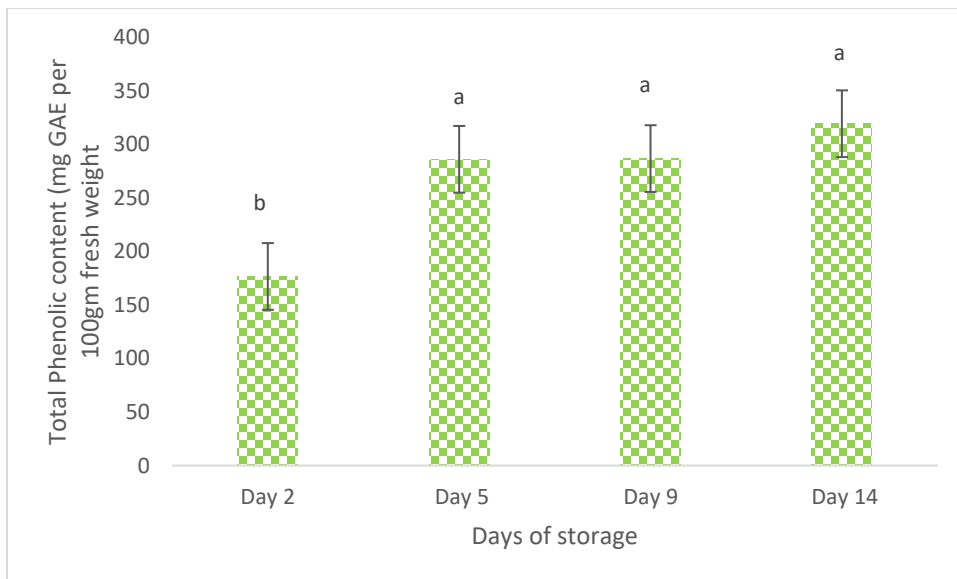
691 LS-means with the same letter are not significantly different



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693 Figure 12: Mean TSS on days of storage in both treated and non-treated strawberries. LS-means
 694 with the same letter are not significantly different

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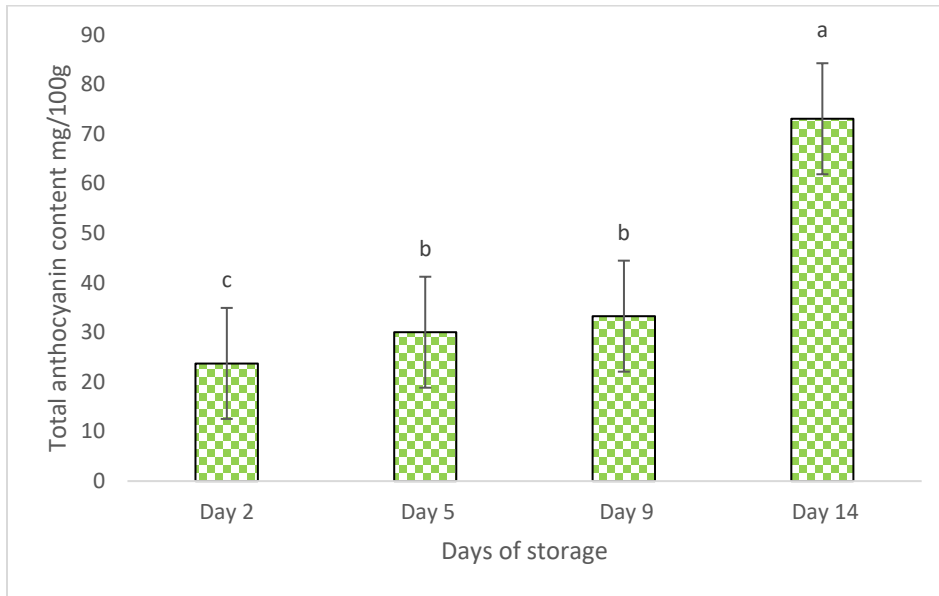
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697 Figure 13: Mean TPC on days of storage in both treated and non-treated strawberries . LS-means
 698 with the same letter are not significantly different

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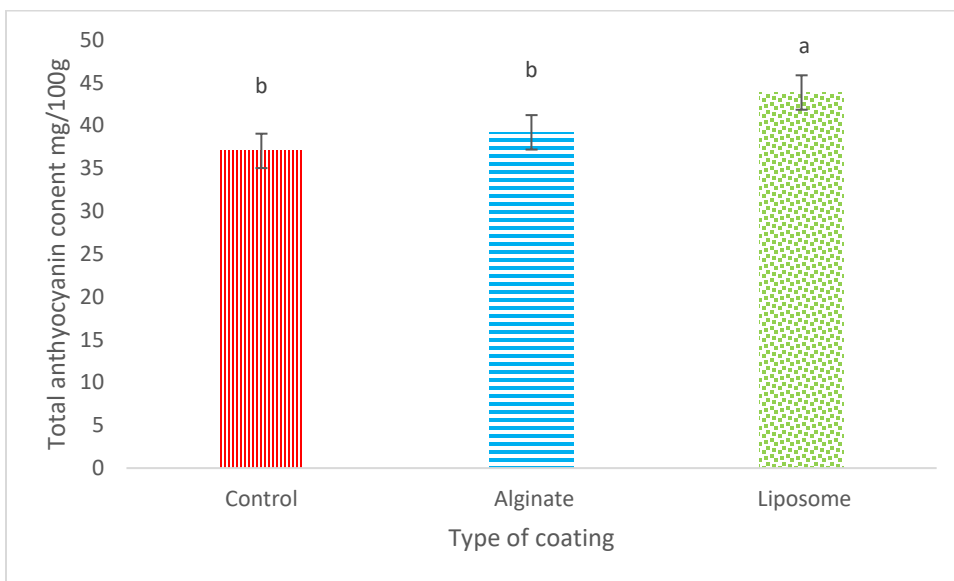
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703 Figure 14: Mean Total anthocyanin on days of storage. LS-means with the same letter are not
704 significantly different

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707 Figure 15: Mean Total anthocyanin on types of coatings. LS-means with the same letter are not
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